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But even the humble covalent bond - the basic building block of chemistry - is a consequence of the quantum interactions of electrons. Once these rules were worked out in the 1920s, scientists realised that, for the first time, they had a detailed theory of how chemistry works.

[Condenser \(laboratory\) - Wikipedia](#)

C 8 H 18 is the formula for octane. The oxygen (O 2) is consumed from the intake air.Nitrogen (N 2) is also present in the atmospheric air, but ideally does not participate in any reactions (it is quite inert at low temperatures).Note that the combustion products are carbon dioxide (CO 2) and water (H 2 O), if the combustion is 'perfect'. Also, note that there are the same number of each kind ...

[17.1. Electrochemical Cells - Chemistry LibreTexts](#)

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